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ಕ್ರಮಾಂಕ/ No.: MU/ACC/CR.7/CBCS-PG(SLB)/2017-18/A2

ಕುಲಸಚಿವರ ಕಛೇರಿ ಮಂಗಳಗಂಗೋತ್ರಿ – 574 199 Office of the Registrar Mangalagangothri – 574 199

ದಿನಾಂಕ/Date: 10.05.2017

NOTIFICATION

Sub: Changes in the title of some courses of I & II semester M.Sc. in Biochemistry degree programme and syllabus thereon.

- Ref: 1) This Office Notification No.: MU/ ACC/ CR.7/ CRCS·PG(SLB)/ 2016-17/ A2, dated 17.08.2016
 - 2) Proceedings of the meeting of BOS in Biochemistry held on 7.09.2016.
 - 3) Approval of the Academic Council at its meeting held on 3.02.2017 vide Agenda No.3:21 (2016-17)

Pursuant to the above, the P.G. Board of Studies in Biochemistry has changed the title of the courses BCH403, BCP407 and BCP459 of M.Sc. in Biochemistry degree programme and framed the syllabus for these courses.

The Academic Council at its meeting held on 3.02.2017 has approved this changes in course titles and syllabus thereon which is hereby notified from the academic year 2017-18 onwards.

REGISTRAR

To:

- 1) The Chairman of the department concerned.
- 2) The Principals of the Colleges concerned.
- 3) The Registrar (Evaluation), Mangalore University.
- 4) The Chairman, P.G. BOS in Biochemistry, Mangalore University.
- 5) The Superintendent (ACC), O/o the Registrar, Mangalore University.
- 6) Guard File.

MANGALORE UNIVERSITY

Department of Studies in Biochemistry, Mangalore University

PG Centre, Chikka Aluvara, Thorenoor Post, Kodagu District, Karnataka, 571 232

PREAMBLE

Revision of syllabi for the two years' Master Degree (Choice Based Credit System- Semester Scheme) Program in Biochemistry.

PG BOS in Biochemistry has revised and prepared the syllabi (CBCS based) for the PG course in Biochemistry

by giving certain guidelines to offer Hard Core, Soft Core and Open Elective courses with credits to each course amounting to 90 credits for the entire program.

There are totally 9 Hard Core theory courses, 3 Hard Core practical courses and one Hard core project work in 4th semester with a total Hard Core credits of 53. In the 4th semester, each student has to take up a research project for which 5 credits are allotted. A total of 9 Soft Core theory courses and 3 Soft Core practical courses with a total of 31 Soft Core credits are being introduced. Board of Studies in Biochemistry has carefully chosen two Open Elective courses for the selection by the students from other disciplines, one each in 2nd and 3rd semester, with total credits of 6. Therefore, grand total credits for the program = **90**.

We have given choice for the soft core courses in the 1^{st} , 2^{nd} and 3^{rd} semesters for Biochemistry post graduates.

A detailed skeleton of the entire program is being tabulated for the benefit of the aspiring post graduates. Syllabi of all the four semesters are prepared and being uploaded.

Other important aspects such as University question paper pattern, internal assessment examinations, allotment of marks and the approximate dates of the internal examinations are being tabulated with a discussion in the Board of Studies in Biochemistry.

C	C	IIl			nistry (CBCS)	TI	T-4-1
S.	Semester	Hard	Soft	Open	Practical,	Theory	Total
No.		core credits	core credits	elective credits	Project*		credits
1.	I Semester	20	3	-	2 (H)	3(H) + 1(S)	23
2.	II Semester	12	9	3	1 (H)	2(H)+4(S)	24
3.	III Semester	12	9	3	2(S)	3(H)+2(S)	24
4.	IV	09	10	-	1(S)	1(H)+2(S)	19
	Semester				1		
					1Project*(H)		
	Total	53	31	6	6 + 1*	18	90

Two-year Master's Degree Course (Four Semesters) M Sc Biochemistry (CBCS)

H = Hard core; S = Soft core

MSc Biochemistry CBCS (All 4 Semesters)

Serial No.	Paper code	Title of the paper
1.	BC H 401	Bioorganic & Biophysical Chemistry
2.	BC H 402	Biomolecules
3.	BC H 403	Analytical Biochemistry
4.	BC P 406	General Biochemistry
5.	BC P 407	Biochemical Techniques
6.	BC H 451	Enzymology
7.	BC H 452	Metabolism of Fuel Molecules
8.	BC P 458	Practical Enzymology
9.	BC H 501	Molecular Biology
10.	BC H 502	Immunology
11.	BC H 503	Metabolism of Nitrogen Containing Compounds
12.	BC H 551	Biotechnology
13.	BC H 555	Project

HARDCORE

SOFTCORE

Serial No.	Paper code	Title of the paper
1.	BC S 404	Human Physiology
2.	BC S 405	General Microbiology
3.	BC S 453	Nutrition
4.	BC S 454	Plant Biochemistry
5.	BC S 455	Clinical Biochemistry
6.	BC S 456	Bioethics & Biosafety
7.	BC P 459	Practical Clinical Biochemistry
8.	BC S 504	Genetics
9.	BC S 505	Nanotechnology
10	BC S 506	Food Science
11.	BC P 508	Molecular Biology & Immunology
12.	BC P 509	Microbiology & Cell Biology
13.	BC S 552	Cell Biology
14.	BC S 553	Bioinformatics & Biostatistics
15.	BC P 554	Practical Biotechnology

OPEN ELECTIVES FOR OTHER DISCIPLINES

Serial No.	Paper code	Title of the paper
1.	BC E 457	Biochemistry in Day-To-Day Life
2.	BC E 507	Health and Disease

FIRST SEMESTER

Serial No.	Paper Code	Title of the paper	Instruction hours /week	Duration of Examination in hours	Marks	Credits	Theory(T)/ Practical(P)	Hardcore(HC) / Soft core(SC)
1.	BC H 401	Bioorganic & Biophysical Chemistry	4	3	70+30=100	4	Т	НС
2.	BC H 402	Bio-molecules	4	3	70+30=100	4	Т	НС
3.	BC H 403	Analytical Biochemistry	4	3	70+30=100	4	Т	НС
4.	BC S 404	Human Physiology*	3	3	70+30=100	3	Т	SC
5.	BC S 405	General Microbiology#						
6.	BC P 406	General Biochemistry	8	6	70+30=100	4	Р	HC
7.	BC P 407	Biochemical Techniques	8	6	70+30=100	4	Р	НС

*# There is a choice between BC S 404 and BC S 405. SECOND SEMESTER

Serial No.	Paper Code	Title of the paper	Instruction hours /week	Duration of Examination in hours	Marks	Credits	Theory(T)/ Practical(P)	Hardcore(HC) / Soft core(SC)
1.	BC H 451	Enzymology	4	3	70+30=100	4	Т	НС
2.	BC H 452	Metabolism of Fuel Molecules	4	3	70+30=100	4	Т	НС
3.	BC S 453	Nutrition*	3	3	70+30=100	3	Т	SC
	BC S 454	Plant Biochemistry#						
4.	BC S 455	Clinical Biochemistry*	3	3	70+30=100	3	Т	SC
7.	BC S 456	Bioethics & Biosafety#						
9.	BC E 457	Biochemistry in Day-To-Day Life	2	3	70+30=100	3	Т	SC
8.	BC P 458	Practical Enzymology	8	6	70+30=100	4	Р	НС
9.	BC P 459	Practical Clinical Biochemistry	8	6	70+30=100	3	Р	SC

*# There is a choice between - 1. BC S 453 and BC S 454. & 2. BC S 455 and BC S 456.

3. BC E 457 is an open elective course for other disciplines.

THIRD SEMESTER

Serial No.	Paper Code	Title of the paper	Instruction hours /week	Duration of Examination in hours	Marks	Credits	Theory(T)/ Practical(P)	Hardcore(HC) / Soft core(SC)
1.	BC H 501	Molecular Biology	4	3	70+30=100	4	Т	НС
2.	BC H 502	Immunology	4	3	70+30=100	4	Т	НС
3.	BC H 503	Metabolism of Nitrogen Containing Compounds	4	3	70+30=100	4	Т	НС
4.	BC S 504	Genetics*	3	3	70+30=100	3	Т	SC
5.	BC S 505	Nanotechnology#						
6.	BC S 506	Food Science \$	3	3	70+30=100	3	Т	SC
7.	BC E 507	Health and Disease	2	3	70+30=100	3	Т	SC
8.	BC P 508	Molecular Biology & Immunology	8	6	70+30=100	3	Р	SC
9.	BC P 509	Microbiology & Cell Biology @	8	6	70+30=100	3	Р	SC

*# There is a choice between BC S 504 and BC S 505 \$@ There is a choice between BC S 506 and BC P 509

FOURTH SEMESTER

Serial No.	Paper Code	Title of the paper	Instruction hours /week	Duration of Examination in hours	Marks	Credits	Theory(T)/ Practical(P)	Hardcore(HC) / Soft core(SC)
1.	BC H 551	Biotechnology	4	3	70+30=100	4	Т	НС
2.	BC S 552	Cell Biology	3	3	70+30=100	3	Т	SC
3.	BC S 553	Bioinformatics & Biostatistics	3	3	70+30=100	3	Т	SC
4.	BC P 554	Practical Biotechnology	8	4	70+30=100	4	Р	SC
5.	BC P 555	Project Work *	10	-	70+30=100	5	Project	НС

*BC P 555 Research project work is compulsory to all students.

University theory question paper pattern

Ser. No.	Question type	Marks
1.	Answer any ten questions out of twelve	2 x 10= 20
2.	Answer any five questions out of eight	$10 \ge 5 = 50$
	Hard core: Two questions from each Unit and the	
	remaining questions from any of the four units for short	
	answers	
	Soft core: Three questions from each unit and the	
	remaining questions from any of the three units for short	
	answers.	

Internal Assessment (Theory & Practical) Examination

Ser. No.	Description	Test	Marks
1.	At the end of 8 th week	C1	30
2.	At the end of 14 th week	C2	30
		C1+C2 /2	Average of two

C1/C2 Theory Marks Allotment

Ser. No.	Description	Marks
1.	Assignment	10
2.	Test	20
	Total	30

Allotment of C1/C2 Practical Marks

Ser. No.	Description	Marks
1.	Practical Test C1	30
2.	Practical Test C2 + Class Seminar	30

I SEMESTER –ANALYTICAL BIOCHEMISTRY: BCH 403 HARD CORE

Total Number of Lecture Hours: 56

Unit I

Preliminary techniques in Biochemistry: Animal and plant models, Investigation with isolated organs and tissues, Introduction to animal and plant cell culture. Investigation with microorganism and their mutant (auxotroph), yeast, *Ceanorhabditis elegans, Arabidopsis thaliana* and *Drosophila melanogaster* as model specimen for biochemical investigations. **Cell fractionation techniques**: Cell lysis, homogenization, extraction, salting in, salting out, dialysis and ultra -filtration. **Centrifugation**: Basic principles of sedimentation, types of centrifuges and rotors. Preparative Centrifugation – Differential and Density gradient, Sub-cellular fractionation, Marker enzyme analysis, Analytical Centrifugation - application and design.

Unit II

Chromatography: Introduction, partition coefficient, Modes of chromatography, liquid and solid phases, paper chromatography and Thin-layer Chromatography (TLC): Principle, procedure and application, Column chromatography: Basic components, selection of stationary and mobile phase, matrices. Adsorption chromatography (hyroxyapatite and Hydrophobic interaction), Partition (normal phase and reverse phase) Ion exchange (Cation and anion exchange), Gel filtration, affinity chromatography, High performance liquid chromatography (HPLC), Fast protein liquid chromatography (GLC).

Unit III

Electrophoretic techniques: Principle, Non-denaturing, denaturing electrophoresis, agarose gel electrophoresis, isoelectric focusing, pulsed field electrophoresis, capillary electrophoresis, Visualising separated components - staining for proteins and nucleic acids, fluorescence, PAS staining, zymogram. **Spectroscopic techniques**: Beer-Lambert's Law and its limitations, Extinction coefficient, Principles & Applications: Colorimeter, UV-Vis Absorption spectroscopy, Fluorescence Spectroscopy, Mass spectrometry, Infrared and Raman Spectroscopy, Nuclear Magnetic Resonance, Electron Spin Resonance, Circular dichroism spectroscopy, X-ray crystallography.

Unit IV

Isotopes in Biochemistry: Isotopes, Types of radioactive decay, Units of radioactivity, Interaction of radioactivity with matter, Detection and measurement of radioactivity: Methods based on gas ionization (Geiger-Muller counter), Excitation (Scintillation counting) and Photographic methods. Specific activity, commonly used isotopes (Tritium, Carbon-14, Phosporous-32, Sulfur-35, Iodine-131), Advantages and restriction of radiotracer experiments, safety aspects, Applications of radioisotopes in biological sciences.

REFERENCES:

- Freifelder D. M. Physical Biochemistry- Application to Biochemistry and Molecular Biology, 2nd ed., W.H. Freeman, 1982.
- 2. Principles and Techniques of Biochemistry and Molecular Biology, ed., Keith Wilson & John Walker, March 2010, Cambridge Univ. Press.
- 3. West & Todd. Biochemistry. 4th ed., Oxford and IBH.
- 4. Upadhyay and Upadhyay. Biophysical Chemistry

Total Number of Credits: 04

14 hrs.

14 hrs.

14 hrs.

14 hrs.

I SEMESTER - GENERAL BIOCHEMISTRY: BC P 406

PRACTICAL - HARD CORE: 4 CREDITS- 8 HOURS/WEEK

- 1. Buffers: a) Introduction b) Preparation
- 2. Quantitative determination of protein concentration by Biuret method.
- 3. Estimation of proteins by Lowry's method.
- 4. Estimation of proteins by Bradford method.
- 5. Bicinchonic acid protein assay.
- 6. Measurement of protein concentration by UV spectroscopy.
- 7. Estimation of glucose by Dinitrosalicylic acid method.
- 8. Estimation of glucose by Anthrone method.
- 9. Estimation of ascorbic acid by DNPH method.
- 10. Estimation of inorganic phosphate by Fiske- Subba Raw's method.

REFERENCES:

- 1. Introduction to practical Biochemistry. David T. Plummer
- 2. Lab Manual of Biochemistry. By Nigam. 2007. Tata McGraw-Hill Education, USA.

I SEMESTER - BIOCHEMICAL TECHNIQUES: BC P 407

PRACTICAL - HARD CORE: 4 CREDITS- 8 HOURS/WEEK

- **1.** Detection of amino acids by circular chromatography
- 2. Detection of amino acids by ascending chromatography.
- 3. Detection of amino acids by descending chromatography.
- 4. Detection of amino acids by 2D- paper chromatography.
- 5. Detection of amino acids by thin layer chromatography.
- 6. Detection of lipids by thin layer chromatography.
- 7. Detection of carbohydrates by paper chromatography.
- 8. Saponification number of oil and fat.
- 9. Iodine number of oil and fat.
- **10.** Acid precipitation of proteins.
- **11.** Preparation of casein from milk and qualitative estimation of proteins.
- 12. Purification of proteins: Ammonium sulphate precipitation (salting out), Dialysis, Ion

exchange, Gel filtration.

- 13. Separation and detection of proteins Native PAGE, Denaturing PAGE, IEF.
- 14. Agarose gel electrophoresis DNA.

REFERENCES:

- 1. Practical Clinical Biochemistry, Harold Varley, Inter science Publishers Inc, 2002
- 2. Clinical Chemistry: Theory, Analysis and Correlation. Kaplan, L.A. and Pesce, A.J., 4th ed. Mosby, 2003.
- 3. Introduction to practical Biochemistry. David T. Plummer
- 4. Nigam. 2007. Lab Manual of Biochemistry. By. Tata McGraw-Hill Education, USA

II SEMESTER - PRACTICAL ENZYMOLOGY: BC P 458

PRACTICAL - HARD CORE: 4 CREDITS- 8 HOURS/WEEK

Salivary Amylase: Activity, Specific activity, Optimum pH and Temperature, pH and Temperature Stability, energy of activation, Km, Vmax, effect of metal ions, Purification by ammonium sulphate fractionation and enzyme characterization.

Assay methods and some characterization of invertase from yeast, acid phosphatase from potato, protease from papaya and esterase from peas.

REFERENCES:

1.Enzymes: A Practical Introduction to Structure, Mechanism, and Data Analysis; Robert A. Copeland, Wiley- VCH Publishers (2000).

- 2. Enzyme Kinetics and Mechanism; Paul F. Cook, W. W. Cleland, Garland Science (2007).
- 3. Biochemical Calculations, Irwin H. Segel (1976) 2nd Ed. John Wiley and Sons.
- 4. Methods in Enzymology; Colowick S.P. et al., Vol. 152, Academic Press, (1987).
- 5. Methods of Enzymatic Analysis; Berg Meyer Vol. 1-X, (1974).

6. Basic Biochemical Laboratory Procedures and Computing, R. Cecil Jack (1995) Oxford University.

7. Enzyme Kinetics; Roberts, D.V. (1977), Cambridge University Press.

II SEMESTER PRACTICAL CLINICAL BIOCHEMISTRY: BC P 459

PRACTICAL - SOFT CORE: 3 CREDITS- 8 HOURS/WEEK

- 1. Urinanalysis Normal and Abnormal.
- 2. Estimation of serum cholesterol by Zak's method.
- 3. Estimation of protein and A-G ratio by biuret method
- 4. Estimation of glucose by Folin- Wu method.
- 5. Estimation of glucose by Somogyi-Shaffer-Hertmann method.
- 6. Estimation of glucose by Hegedson Jenson method
- 7. Estimation of Iron by Wong's method
- 8. Estimation of Serum SGOT, SGPT, LDH, ALP using kits.
- 9. Estimation of Urea, uric acid, creatinine using kits.
- 10. Estimation of TAG, Cholesterol using kits. Determination of HDL and LDL cholesterol.
- 11. Electrophoresis of hemoglobin and isoenzymes.

REFERENCES:

- 1. Tietz Fundamentals of Clinical Chemistry and Molecular Diagnostics, Carl A. Burtis, David E. Bruns. 7th ed. Elsevier, 2014.
- 2. Practical Clinical Biochemistry, Harold Varley, Interscience Publishers Inc, 2002
- 3. Clinical Chemistry: Theory, Analysis and Correlation. Kaplan, L.A. and Pesce, A.J., 4th ed. Mosby, 2003.
- 4. Introduction to Practical Biochemistry. David T. Plummer
- 5. Lab Manual of Biochemistry. By Nigam. 2007. Tata McGraw-Hill Education, USA.