

Department of Industrial Chemistry

ICP 408: PHYSICAL CHEMISTRY PRACTICALS-I

Course Outcomes:

- Students learn following technique under physical chemistry.
- Determining physical constants using refractometry, adsorption experiment and viscometry, Study on kinetics of hydrolysis, catalytic effect and calculation of thermodynamic parameters.

Any 12 experiments are to be carried out

- 1. Analysis of a binary mixture and determination of molar refraction of a solid and the composition of chloroform and acetone in its azeotropic mixture by refractometry.
- 2. Analysis of a binary mixture of two miscible liquids by viscometry and the relation between viscosity of a solution and the electrical conductivity.
- 3. Study of variation of viscosity of a liquid with temperature.
- 4. Determination of parachor value for CH₂ group by S.T method, the composition of a solution by S.T measurement and the CMC of a soap solution by S.T measurement.
- 5. Surface tension concentration correlation for solutions (Gibbs equation).
- 6. Verification of F& L adsorption isotherms for acetic &oxalic acids on activated charcoal.
- 7. Analysis of a binary mixture by surface tension method.
- 8. Adsorption of iodine on charcoal from alcoholic solution.
- 9. Study of adsorption of picric acid on charcoal using a calorimeter.
- 10. Acid catalysed hydrolysis of methyl acetate and determination of catalytic strength of an acid.
- 11. Saponification of ethyl acetate by conductivity method.
- 12. Reaction between potassium persulphate and potassium iodide (including the study of salt effect and catalysis by Ag ⁺, Fe ²⁺ and Cu ²⁺ ions).
- 13. Decomposition of diacetone alcohol by NaOH & Hydrolysis of t-Butylchloride.
- 14. Reaction between hydrogen peroxide and HI.
- 15. Determination of solubility of lead iodide at different T & hence molar heat of solution.
- 16. Determination of heat of solution of a sparingly soluble solute.
- 17. Any other interesting experiments

References

- 1. Findlay's Practical Physical Chemistry, B. P. Levitt, Longman, London.
- 2. Experiments in Physical Chemistry, James and Prichard.
- 3. Experimental Physical Chemistry, Daniels et al.
- 6. Experimental Physical Chemistry, Das & Behera, Tata McGraw Hill, New Delhi, 1983.
- 7. Advanced Practical Physical Chemistry, Yadav, 1989.
- 8. Experiments in Physical Chemistry, J.C.Ghosh, Bharathi Bhavan, 1974.