UNIT -II :

[12 Hours]

Dyes: Introduction, modern theories of colour and chemical constitution. Classification of dyes, methods of applying dyes to the fabrics. A general study of Azo dyes- Orange –II, rosanthrene O, Naphthol blue black 6B, Mordant brown, Congo red, Methyl orange, Chrysoidin G, Bismark brown.

Triphenylmethane dyes- Malachite green, Rosaniline, Crystal violet and Phenolphthalein;

Cyanin dyes- Ethyl Red, Cyanin blue and Quinaldine, Reactive dyes and Optical brighteners-

Tinapal and Blankophor.

Pigments: Fast violet, Lake red and Orange R.

UNIT - III :

Insecticides: Introduction, classification, mode of action and synthesis of Methoxychlor,chlordane, heptachlor, Hexachlorocyclohexane, Parathion, Diazenon, Sevin and Beygon. Naturally occurring insecticides-pyrethroids-natural pyrethrins-isolation and structures, synthetic pyrethroids.

Insect Pheromones: Introduction, use in insect pest control. Synthesis of disparlure, grandisoland bomykol.

Fungicides: Introduction, Systemic fungicides-types & examples.

Herbicides: Introduction, study of sulfonyl ureas and heterocyclic sulphonamides.

Fumigants and repellants. Mechanism of action and toxicities of insecticides, fungicides and herbicides.

References

1. Polymer Science- V.R.Gowariker, N.V.Vishwanathan&T.Shridhar (Wiley Eastern) 2008.

2. Textbook of Polymer Science, 3rd Edition, Fred W. Billmeyer(Wiley) 1984.

3. A Textbook of Synthetic Dyes- O.D. Tyagi&M. Yadav(Anmol Publications) 2002.

4. <u>Textbook of Dyes - A. Arora(SonaliPublications)</u> 2009.

5. Synthetic Dyes – Vol-I – Venkataraman, 1999.

6. Synthesis and Chemistry of Agrochemicals, Vol I & II, ACS, Wahington

7. Chemicals for Crop Protection and Pest Managements, M B Green, G.S. Hartley

West, Pergamon.

8. Chemistry of Insecticides and Fungicides, SreeRamulu, Oxford & IBH, 1985.

AC S 555: APPLIED ELECTROCHEMISTRY

COURSE OUTCOME:

- The course covers important practical applications of electrochemistry. Batteries, fuel cells, sensors and electroplating techniques are dealt with.
- The use of electrochemical techniques in environmental related issues are discussed
- chemical processes such as costing and design of electrochemical processes,
- They can learn important organic and inorganic reactions which can be carried out in industries and modern technological developments in electrochemical industrial processes.

UNIT-I: [12 Hours] Electrochemical Energy System: Electricity storage-Importance, storage density, Fundamentalsand classification of batteries, Primary battery (Laclanche-dry cell and Alkaline cell). Secondary battery (acid and alkaline). Reserve batteries. Lithium batteries - (primary and secondary and lithium based conducting polymer battery). Fuel cells – introduction, classification, H2-O2 and bio-cells. 5hrs

[12 Hours]

Bio-electrochemistry- Introduction, Membrane potential - theoretical and modern approach.Electrical conduction in biological organism, Electrochemical communication in biological organisms. **3hrs**

Sensors: Biosensors: Introduction electrochemical bio-sensors- characteristics, use as atransducer, types. **Ion-Sensors**: Ion-selective electrode: Introduction, Types. Analytical and biological applications of sensors. 4hrs

UNIT-II:

[12 Hours]

Metallurgical Processing: Electroplating-fundamentals, mechanism of electrodeposition of metals, application of electroplating. Brief account of Electroless plating, Conversion coatings, Electrophoretic painting.

Metals and materials processing-theory and applications of Electroforming

andElectrochemical etching. Production of metals by electro winning and electrorefining. **Electrochemistry of Environment**: Introduction, Global warming. Electrochemistry in transport system, fixing of CO₂, sewage disposal, treatment of waste, Metal ion removal and metal recovery. Treatment of liquors containing dissolved chromium.

UNIT-III:

[12 Hours]

Electrochemical Engineering: General considerations, costing and technology of electrolyticprocess, electrolysis parameters, principles of cell design, laboratory data and scaleup, performance and figures of merit. 4hrs **Industrial Electrochemistry**: Fundamentals, electro- organic synthesis (Kolbes synthesis, oxidation and reduction of hydrocarbons, reduction of nitro-compounds); Electro inorganic synthesis of fluorine and ozone. Synthesis of metal salts via anodic dissolution 4hrs.

Industrial Application- A Case study:- The chlor-alkaly industry: Introduction, Generalconcepts of brine elecrolysis, modern technological developments (electrode materials, membrane), chlorine cell technologies (diaphragm cells, membrane cell). 4hr

References:

1. Modern Electrochemistry, 2nd Ed. Vol.1,2A &2B, Bockris& Reddy (Plenum, NY) 1998

2. Chemical & Electrochemical Energy Systems, R. Narayan & B. Viswanathan (University Press), 1998.

3. Industrial Electrochemistry, D. Peltcher& F. C. Walsh (Chapman & Hall)1990.

4. Biosensors-theory and Applications, Donald G. Burek, (Technomic), 1993.

5. Principles and Applications of Electrochemistry–Crow (Chapman hall, New York) 2014

6. Fundamentals of Electrochemistry, Fulkner and A. J. Bard, Wiley India, 2006.

AC S 556: REACTION KINETICS & NUCLEAR CHEMISTRY COURSE OUTCOME:

- The course content consists of two topics, reaction kinetics and radiation chemistry. The first part covers principles of various modern techniques useful for study of reaction rates.
- It covers the applications of kinetics in understanding mechanism of some important organic and inorganic reactions.
- It includes the theoretical aspects of evaluation of energy of activation of reactions. In the second part, the nuclear reactions, radiation measurements,
- They also learn design and functioning of nuclear reactors are taught.