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**BSCPHC 357**

**Credit Based VI Semester B.Sc. Degree Examination, September 2022  
(2020 – 21 and Earlier Batches)  
PHYSICS (Paper – VII)  
Nuclear Physics**

Time : 3 Hours

Max. Marks : 80

- Instructions :** 1) Answer questions from **all** Units.  
2) Multiple choice questions must be answered in the **first two** pages of the answer book.  
3) Scientific calculators are **allowed**.

**PART – A**

1. Answer the following questions by choosing the most appropriate answer. **(1×10=10)**

i) The heaviest stable nucleus is

- a)  ${}_{209}\text{Bi}$                       b)  ${}_{208}\text{Pb}$                       c)  ${}_{207}\text{Pb}$                       d)  ${}_{206}\text{Pb}$

ii)  $\beta$  particle is an electron of \_\_\_\_\_ origin.

- a) Atomic    b) Nuclear  
c) Both of the above                              d) None of these

iii) Condition for secular equilibrium is

- a) Half-life of parent is equal to that of daughter element  
b) Half-life of parent is less than that of daughter element  
c) Half-life of parent is slightly greater than that of daughter  
d) Half-life of parent is very large compared to that of daughter

iv) Which of the following form the pair of mirror nuclei ?

- a)  ${}_{6}\text{C}^{11}$  and  ${}_{5}\text{B}^{11}$     b)  ${}_{3}\text{Li}^7$  and  ${}_{4}\text{Be}^7$   
c)  ${}_{1}\text{H}^3$  and  ${}_{2}\text{He}^3$     d) All of these

v) The nuclear quadrupole moment of nuclei containing magic number of nucleons is

- a) Positive    b) Negative  
c) Zero    d) None of these

P.T.O.



- vi) If 5 gram of U-235 is completely destroyed in a reactor, the energy released would be  
a)  $45 \times 10^3$  J      b)  $15 \times 10^{10}$  J      c)  $45 \times 10^{17}$  J      d)  $60 \times 10^{15}$  J
- vii) Nuclear reactor at Kaiga is  
a) Research reactor      b) Fusion reactor  
c) Breeder reactor      d) Power reactor
- viii) Neutrino belongs to the family called  
a) Hadrons      b) Leptons  
c) Baryons      d) Hyperons
- ix) The maximum energy gained by the electron in a betatron is  
a)  $B e r C$       b)  $B e r^2 C$   
c)  $B^2 e r C$       d)  $B e r C^2$
- x) Cosmic rays were discovered by  
a) Thomson      b) Curie      c) Hess      d) Millikan

2. Answer **any five** of the following : **(2×5=10)**

- i) Give the empirical relation between range and velocity of an alpha particle.
- ii) According to Pauli's theory how many types of neutrinos are there ? What is the difference between them ?
- iii) Show that nuclear density is constant.
- iv) Which has higher binding energy  ${}_1\text{H}^3$  or  ${}_2\text{He}^3$  ? Comment on the difference.
- v) How are thermal neutrons produced ?
- vi) Name the four basic interactions of the nature.
- vii) Draw GM characteristics.

PART – B

Unit – I

3. a) State and explain Geiger-Nuttal law.
- b) With elements A, B and C forming a radioactive series (C being stable), derive an expression for the number of atoms of B if at start B was not present in the sample. Hence derive the condition for secular equilibrium. **(4+6)**

OR



4. a) What is radioactive dating ? Explain radio carbon dating.  
b) Explain 3 types of Beta decay, using Paulis neutrino hypothesis. What are the condition for the emission of  $\beta^+$  and electron capture ? **(4+6)**
5. a) Radon the disintegration product of radium is in equilibrium with 1 gram of radium. Find the mass of radon. Half-life of Ra-226 = 1590 years, Rn-222 = 3.82 days. **5**

OR

- b) 1 gram of radium is reduced by 2.1 mg in 5 years by alpha decay. Calculate the half-life period of radium. **5**

### Unit – II

6. a) Explain how shell model accounts for the first 3 magic numbers.  
b) Derive  $\alpha$  particles scattering formula assuming expression for impact parameter. Also mention the factors on which the number of  $\alpha$ -particles scattered depend. **(4+6)**

OR

7. a) Using Heisenberg's uncertainty principle estimate the rest mass of meson assuming the range of nuclear forces as 1.5 fm.  
b) What is a mass spectrograph ? Describe with theory Dumpster's mass spectrograph. Explain how isotopic abundances can be determined. **(4+6)**
8. a) In a Dumpster's mass spectrograph a 1000 V accelerating potential brings singly ionized Mg – 24 on the slit. Calculate the potential difference to bring singly ionized Mg – 25 on the slit, the magnetic field being kept constant. **5**

OR

- b) In Rutherford  $\alpha$ -ray scattering experiment, the flux of  $\alpha$ -particles observed at 100 is 106 per minute. Calculate the flux of  $\alpha$ -particles observed at (i)  $90^\circ$  (ii)  $180^\circ$ . **5**

### Unit – III

9. a) Write a short note on fast breeder reactors.  
b) Explain nuclear fission with an example. Obtain the four factor formula for thermal reactors and discuss the condition for criticality. **(4+6)**

OR



10. a) Give Rutherford's experiment on artificial transmutation.
- b) With a diagram explain working of a nuclear reactor. What is the difference between pressurised water reactor and boiling water reactor ? **(4+6)**
11. a) It is proposed to produce 100 MW of electrical power on an average in a nuclear reactor having 20% efficiency, using U-235. Calculate the amount of U-235 required per day for continuous operation. Given energy released per fission of U-235 is 200 MeV. **5**

OR

- b) Show by mass-energy calculation whether the following reactions are endoergic or exoergic.  $^{14}\text{N} (\alpha, p) ^{17}\text{O}$ . **5**

#### Unit – IV

12. a) Explain the working of a semiconductor detector.
- b) Explain the classification of fundamental particle with respect to mass, spin and interaction. **(4+6)**
13. a) Explain Van Allen radiation belts.
- b) With neat diagram explain the working of linear accelerator. **(4+6)**
14. a) A cyclotron accelerate protons to 4 MeV. To what energy will the cyclotron accelerate (a) alpha particles and (b) deuteron. **5**

OR

- b) In a certain betatron the maximum magnetic field at orbit was  $0.4 \text{ Weber/m}^2$ , operating at 50 Hz with a stable orbit diameter of 1.524 m. Calculate the average energy gained per revolution and the final energy of the electrons. **5**
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